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Assessment of oxidative changes in tobacco seed oil stored at different conditions

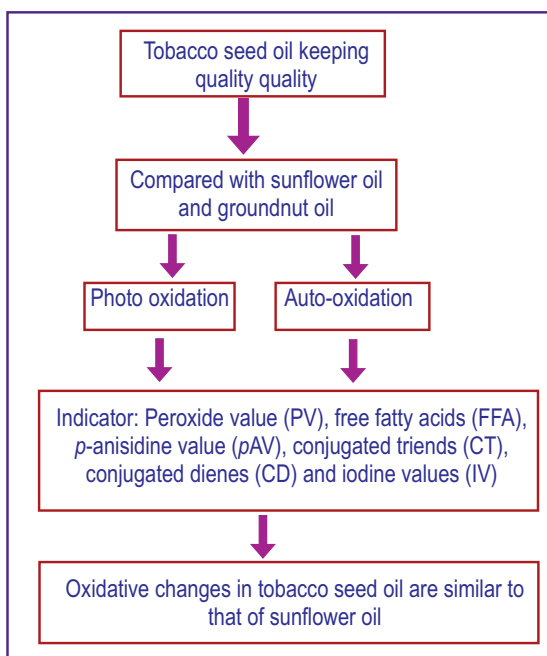
Abstract

Aim: Alternative uses of tobacco have gained importance in recent times to sustain crop for non-conventional and economically viable application in food and industries. One of the alternative promising use of tobacco is seed oil having nutritive, pharmaceutical and industrial utility.

Methodology: Experiments were conducted to study the extent of oxidative alterations in tobacco seed oil subjected to ambient and sunlight storage, for a period of 90 days and compared with sunflower and groundnut oils. The magnitude of oxidative changes was monitored by periodical measurement of peroxide value (PV), free fatty acids (FFA), p-anisidine value (pAV), conjugated trienes (CT), conjugated dienes (CD) and iodine values (IV).

Results: Peroxide values significantly increased from 15 days of storage (1.13 meq kg^{-1}) reaching maximum value (8.45 meq kg^{-1}) at 90 days. Photo-oxidation (4.82 meq kg^{-1}) was significantly higher than auto-oxidation (3.80 meq kg^{-1}). The peroxides were significantly different among the three oils with the highest in tobacco seed oil (5.06 meq kg^{-1}) followed by sunflower oil (4.29 meq kg^{-1}) and groundnut oil (3.58 meq kg^{-1}). The pAV increased significantly with increase in days of storage attaining maximum value of 13.84 at 90 days. pAV in tobacco seed oil (7.27) and sunflower oil (7.18) were at a par and were significantly higher than groundnut oil (5.36). Tobacco oil at 90 days of storage showed 2.9% and 7.19% higher content of CD over sunflower oil in photo and auto-oxidations respectively. FFA content in tobacco seed oil (0.153%) and sunflower oil (0.150%) were at a par and were significantly higher than groundnut oil.

Interpretation: The oxidative changes in tobacco seed oil showed nearly similar trends with that of sunflower oil. The results showed that even though tobacco seed oil contains higher levels of unsaturated fatty acids, tobacco oil can be stored like any other edible oils.



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Introduction

Tobacco is a leading commercial crop in India and is grown in 0.45 M ha area with 750 M kg leaf production. However, the anti-tobacco campaign for traditional form of consumption is posing a serious challenge to tobacco production, trade and industry. The economic life of millions of people, including six million farmers, depends on tobacco necessitating the crop to be sustained for its potential alternative uses. The promising alternative use of tobacco is seed oil having nutritive, pharmaceutical and industrial uses (Awolola *et al.*, 2010). The research on alternative uses of tobacco is the order of the day, leading to critical examination of potentials of tobacco as 'an oil seed crop'. An estimated 1300-1500 tonnes of tobacco seed oil is expelled and exported from India to other countries for utilization in paint industry (Deo Singh and Narasimha Rao, 2005). It is used as raw material for preparation of printing inks, dyes (Zlatanova *et al.*, 2007), potential use in food and coating industries (Muktha *et al.*, 2007), production of soaps, shoe polish, varnishes (Purseglove, 1991) and an alternative to diesel fuel (Giannelos *et al.*, 2002). However, tobacco seed oil finds extensive use in paints, varnishes, lubricants and soap industries. There has been a significant gap between demand and supply of edible oil because of limited oil seeds and shifting of acreage to other crops, thus making need to search for alternative sources of edible oils.

Tobacco seed contains 35% oil and linoleic acid is the major fatty acid (66 – 76%) (Siva Raju *et al.*, 2011). Tobacco seed oil shows promise as edible oil due to demonstration of the dietary effect of linoleic acid rich oils like corn or safflower oils in lowering the serum cholesterol (Zlatanova *et al.*, 2007). Different variables involved in oil shelf-life, such as processing, storage conditions, light exposure, type of packaging material, availability of oxygen, and addition of antioxidants affect the quality of vegetable oils/fats. Oxidation of oil may be the primary cause of deterioration in quality, and peroxidation is considered a principal mean of deterioration in oil quality. Peroxidation imparts rancid and undesirable flavors to fat products (Pezzuto and Park, 2002). Warm ambient temperature encourage the onset of oxidation in a silent way. Auto-oxidation proceeds through free radical chain reaction by the attacking the double bond at room temperature. Photo-oxidation on the other hand is a much faster reaction that involves attack at double bond by a singlet 1O_2 species (Eunok Choe and David, 2006). Photo-oxidation postulates that double bond within a fatty acid molecule may be capable of capturing outside source of energy, such as heat and light, to reach a critical excitation level (Fekamrhobo and Obomanu, 2009) and the double bond may break giving rise to a free radical species, which may in turn generate more free radicals. Auto and photo-oxidation, which are natural oxidation and chemical degradation processes of edible oils, also results in rancidity of food items, resulting these processes to convert fatty acid esters of oils into free fatty acids. The present study was undertaken with an objective to evaluate the extent of oxidative deterioration and

shelf-life of tobacco seed oil stored under ambient and sunlight conditions and compared with edible sunflower and groundnut oils.

Materials and Methods

Tobacco seed, sunflower and groundnut oils were extracted by Soxhelt method using hexane and decolorized by Fullers earth. Five grams of seed was pounded well using mortar and pestle and packed into Whatman thimble. A piece cotton was placed at the top to evenly distribute the solvent as it dropped on sample during extraction. Thimbles were placed in butt tubes of the Soxhlet extraction apparatus. The extraction was done with 150 ml hexane for 12 hrs by gentle heating. After extraction, flasks were cooled and hexane was removed using flash evaporator (vacuum evaporator). The flasks were dried in the oven to remove traces of hexane at 70°C and the flasks were cooled in the desiccators and weighed. The process was repeated till constant weight of oil was obtained. The samples were stored in 250 ml translucent PET bottles under controlled setup. Two sets of each samples were separately subjected to auto-oxidation (at ambient temperature in dark) and photo-oxidation (exposure to sun light daily for 7 hours) for a period of 90 days. Analysis was carried out at 15 days interval. Oil samples were analyzed for peroxide value, iodine value, (Cox and Pearson, 1962) and free fatty acids (Lowry and Tinsley, 1976). Conjugated dienes and trienes, in terms of specific extinctions at 232 and 268 nm were determined by spectrophotometer. Oil samples were diluted with isooctane to bring the absorbance within limits (0.2-0.8) and $\epsilon^{1\% 1\text{ cm}}$ was calculated following IUPAC method (IUPAC, 1987). Determination of *p*-anisidine value of the oil samples was determined following IUPAC method (IUPAC, 1987). Samples dissolved in isooctane were allowed to react with *p*-anisidine solution in acetic acid (0.25% w/v) for 10 min to produce a colored complex, and the absorbance value was noted at 350 nm by spectrophotometer. Statistical analysis was done by SAS 9.3 software.

Results and Discussion

Peroxide value (PV) is a sensitive indicator of early stages of oxidative deterioration of fats and oils. The analysis of variance for storage period, oxidation methods and oils was significant for peroxide value (PV). The co-efficient of variation for PV was 8.43 which was reasonably low and correlation co-efficient was 0.99 (Table 1). The storage periods, 0 and 15 days were at a par whereas others were significant. Peroxide values significantly increased from 15 days of storage (1.13 meq kg⁻¹) reaching a maximum value of 8.45 meq kg⁻¹ at 90 days (Table 2). Photo-oxidation (4.82 meq kg⁻¹) was significantly higher than auto-oxidation (3.80 meq kg⁻¹). The peroxide values were significantly different among the three oils with highest in tobacco seed oil (5.06 meq kg⁻¹) followed by sunflower oil (4.29 meq kg⁻¹) and groundnut oil (3.58 meq kg⁻¹) (Table 2), respectively.

Table 1 : Effect of storage period, oxidation methods on chemical properties of tobacco, sunflower and groundnut oils

Source	DF	Peroxide value (meq kg ⁻¹)	p-Anisidine value	Conjugated trienes (ε ^{-1%} λ.268)	Conjugated dienes (ε ^{-1%} λ.232)	Free fatty acids (%)	Iodine value (g l 100 g ⁻¹)
Replications	2	0.8352	0.2317	0.0117	0.289	0.0002	2.1666
Treatments	43	25.961*	63.663*	32.396*	72.967*	0.0149*	699.903*
Days of storage (f1)	6	159.995*	412.73*	217.316*	489.186*	0.0989*	290.013*
Oxidation methods (f2)	1	33.120*	68.775*	27.188*	45.780*	0.0126*	45.841
Oils (f3)	2	22.817*	48.768*	10.055*	33.666*	0.0063*	14089.7*
F1.f2	6	7.216*	7.56*	3.706*	4.3748*	0.0018*	7.8966
F1.f3	12	1.985*	2.174*	1.172*	4.729*	0.0006	4.5251
F2.f3	2	0.670	5.823*	0.5264	1.001	0.0004	2.627
F1.f2.f3	12	0.621	0.939	0.373*	0.3143	0.0001	1.599
Error	82	0.132	0.3289	0.0888	0.2392	0.0002	10.792
C.V.(%)		8.43	8.68	10.64	9.09	9.58	2.87
R ²		0.990	0.990	0.9948	0.9938	0.9761	0.971

f1: Storage period; f2: methods of oxidation; f3: oils; *Significant

Table 2 : Interaction effects of storage period, oxidation methods and oil types on chemical properties

Days of storage (f1)	Peroxide value (meq kg ⁻¹)	p-Anisidine value	Conjugated trienes (ε ^{-1%} λ.268)	Conjugated dienes (ε ^{-1%} λ.232)	Free fatty acids (%)	Iodine value (g l 100 g ⁻¹)
0	0.7022 ^a	0.6133 ^a	0.020 ^a	0.0255 ^a	0.0794 ^a	118.777 ^a
15	1.1300 ^a	1.7767 ^b	0.1267 ^a	1.055 ^b	0.0839 ^a	117.888 ^a
30	2.6333 ^b	4.3233 ^c	0.3100 ^a	1.5750 ^b	0.0933 ^{ab}	116.555 ^a
45	4.3133 ^c	6.7233 ^d	1.1033 ^b	4.0805 ^c	0.1122 ^{ab}	115.222 ^{ab}
60	5.7955 ^d	8.0333 ^e	2.9022 ^c	6.2711 ^d	0.1617 ^c	112.722 ^b
75	7.1400 ^e	10.9333 ^f	6.3750 ^d	11.2644 ^e	0.2011 ^d	110.722 ^b
90	8.4544 ^f	13.8417 ^g	8.7744 ^e	13.3872 ^f	0.2783 ^e	107.777 ^b
Types of oxidation (f2)						
Auto oxidation	3.7971 ^a	5.8676 ^a	2.3371 ^a	4.7778 ^a	0.1343 ^a	114.841 ^a
Photo oxidation	4.8225 ^b	7.3452 ^b	3.2662 ^b	5.9833 ^b	0.1543 ^b	113.634 ^a
Oils (f3)						
Tobacco seed oil	5.0557 ^a	7.2748 ^a	3.2421 ^a	6.1974 ^a	0.1526 ^a	129.785 ^a
Sunflower oil	4.2919 ^b	7.1812 ^a	2.8878 ^b	5.5209 ^b	0.1500 ^a	118.880 ^b
Groundnut oil	3.5819 ^c	5.3633 ^b	2.2750 ^c	4.4233 ^c	0.1302 ^b	94.0476 ^c

Means having different superscripts within a column are significantly different at 1%

Interaction between days of storage and method of oxidation was significant. The peroxide values for two methods of oxidation for 0, 15, 30 and 45 days of storage were at a par. It was observed that photo-oxidation was significantly higher over auto-oxidation from 60 days onwards and reaching a maximum of 9.80 for photo-oxidation at 90 days (Table 3). Interaction between days of storage and oils was highly significant. The peroxide values of three oils at 0 and 15 days of storage were at a par (Table 4). Tobacco seed oil and sunflower oil were at a par for 60, 75 and 90 days of storage and were significantly higher than groundnut oil (Table 4). Interaction between oils and oxidation methods was significant. The peroxide values of three oils were significantly different in auto-oxidation and tobacco seed oil had significantly higher values (4.678 meq kg⁻¹) followed by sunflower (3.66 meq kg⁻¹) and groundnut oil (3.049 meq kg⁻¹). Photo-oxidation of tobacco seed oil and sunflower oil were at a par and significantly

higher than groundnut oil (Table 5). Increase in PV on storage has been reported in sunflower oil and canola oil (Neff *et al.*, 1994) and a higher rate of formation of primary oxidation products in light exposed oils (Khan and Shahidi, 2000).

Analysis of variance for p-Anisidine value (pAV) explained 99% variation with 8.7% co-efficient of variation. ANOVA for all factors and their interactions were significant at 1% level (Table 1). The pAV increased significantly with increase in storage period and reaching maximum of 13.84 at 90 days. pAV in photo-oxidation (7.34) was significantly higher than auto-oxidation (5.87). pAV in tobacco seed oil (7.27) and sunflower oil (7.18) were at par and were significantly higher than groundnut oil (5.36) (Table 2).

Interaction between the storage periods and oxidation methods was significant. Photo-oxidation was significantly higher

Table 3 : Interaction of storage period (f1) and oxidation methods (f2) on chemical properties of tobacco, sunflower and groundnut oils

Factors	Peroxide value (meq kg ⁻¹)	p-Anisidine value	Conjugated trienes (ε ^{-1%} λ.268)	Conjugated dienes (ε ^{-1%} λ.232)	Free fatty acids (%)	Iodine value (g l 100 g ⁻¹)
0dxAO	0.7000 ^h	0.6133 ^l	0.0200 ^h	0.0255 ^l	0.0777 ^h	118.55 ^f
0dxPO	0.7044 ^h	0.6133 ^l	0.0200 ^h	0.0255 ^l	0.0811 ^{gh}	119.00 ^f
15dxAO	1.0800 ^h	1.5666 ^l	0.1200 ^h	1.0511 ^h	0.0811 ^{gh}	117.77 ^f
15dxPO	1.1800 ^h	1.9866 ^l	0.1333 ^h	1.0600 ^h	0.0866 ^{gh}	118.00 ^f
30dxAO	2.8533 ^g	3.6000 ^h	0.1533 ^h	1.2300 ^{gh}	0.0911 ^{gh}	117.00 ^{ef}
30dxPO	2.4133 ^g	5.0466 ^g	0.4667 ^{gh}	1.9200 ^g	0.0955 ^{gh}	116.11 ^{def}
45dxAO	4.0533 ^f	6.2066 ^f	0.8000 ^g	3.3377 ^f	0.1022 ^{fg}	115.55 ^{def}
45dxPO	4.5733 ^{ef}	7.2400 ^e	1.4067 ^f	4.8322 ^e	0.1222 ^f	114.88 ^{cdef}
60dxAO	4.6933 ^e	7.0000 ^{ef}	2.1333 ^e	5.4400 ^e	0.1511 ^e	113.88 ^{bodef}
60dxPO	6.8977 ^b	9.0600 ^d	3.6711 ^d	7.1022 ^d	0.1722 ^{de}	111.55 ^{abcd}
75dxAO	6.0933 ^d	10.226 ^c	5.3133 ^c	10.18 ^c	0.1888 ^d	111.88 ^{bode}
75dxPO	8.1866 ^c	11.640 ^b	7.4367 ^b	12.34 ^b	0.2133 ^c	109.55 ^{abc}
90dxAO	7.1066 ^b	11.853 ^b	7.8200 ^b	12.18 ^b	0.477 ^b	109.22 ^{ab}
90dxPO	9.8022 ^a	15.8300 ^a	9.7200 ^a	14.59 ^a	0.3088 ^a	106.33 ^a

d: days; AO: Auto-oxidation; PO: Photo-oxidation; Means having different superscripts within a column are significantly different at 1%

Table 4 : Interaction of storage period (f1) and oil types (f3) on chemical properties during storage

Factors	Peroxide value (meq kg ⁻¹)	p-Anisidine value	Conjugated trienes (ε ^{-1%} λ.268)	Conjugated dienes (ε ^{-1%} λ.232)	Free fatty acids (%)	Iodine value (g l 100 g ⁻¹)
0dxTO	0.613 ^l	0.946 ^l	0.02 ^e	0.040 ^l	0.081 ^{fg}	134.8 ^c
0dxSO	0.833 ^k	0.653 ^k	0.02 ^e	0.026 ^l	0.081 ^{fg}	123.8 ^{efgh}
0dxGO	0.660 ^l	0.240 ^l	0.02 ^e	0.010 ^l	0.075 ^g	97.66 ^a
15dxTO	1.25 ^{jk}	2.010 ^h	0.16 ^e	1.250 ^l	0.088 ^{efg}	132.6 ^{cd}
15dxSO	1.160 ^k	2.180 ^{hk}	0.12 ^e	0.980 ^l	0.085 ^{fgi}	123.0 ^{ghni}
15dxGO	0.980 ^l	1.140 ^{jk}	0.10 ^e	0.936 ^l	0.078 ^g	98.00 ^a
30dxTO	3.736 ^f	4.946 ^f	0.615 ^{eg}	1.591 ^l	0.098 ^{efg}	132.5 ^{cd}
30dxSO	2.260 ^{hi}	4.660 ^f	0.170 ^e	1.670 ^{hi}	0.093 ^{efg}	121.5 ^{ghij}
30dxGO	1.903 ^l	3.360 ^l	0.145 ^e	1.460 ^l	0.088 ^{efg}	95.66 ^{ab}
45dxTO	5.950 ^d	7.580 ^{eg}	1.170 ^c	5.175 ^f	0.115 ^{de}	130.3 ^{cd}
45dxSO	4.210 ^{jl}	7.270 ^{eg}	0.990 ^g	4.413 ^f	0.113 ^{de}	119.6 ^{hij}
45dxGO	2.780 ^g	5.32 ^{ef}	0.610 ^{eg}	2.666 ^h	0.108 ^{def}	95.66 ^{ab}
60dxTO	6.840 ^h	8.31 ^b	3.546 ^h	7.210 ^g	0.173 ^b	129.0 ^{cdef}
60dxSO	5.890 ^d	9.400 ^b	3.040 ^h	6.293 ^{eg}	0.175 ^b	116.8 ^{jk}
60dxGO	4.6566 ^g	6.390 ^{ab}	2.120 ^c	5.400 ^{ef}	0.136 ^d	92.33 ^{ab}
75dxTO	7.890 ^{ab}	12.060 ^a	7.230 ^{af}	13.330 ^d	0.213 ^b	126.83 ^{defg}
75dxSO	7.286 ^{bh}	11.580 ^a	6.725 ^f	11.113 ^a	0.211 ^h	115.5 ^{jk}
75dxGO	6.243 ^{cd}	8.890 ^b	5.170 ^b	9.350 ^b	0.178 ^b	89.83 ^b
90dxTO	9.110 ^f	15.070 ^c	9.413 ^d	14.875 ^c	0.298 ^c	122.3 ^{ghij}
90dxSO	8.403 ^{ef}	14.255 ^c	9.150 ^d	14.150 ^{cd}	0.290 ^c	111.8 ^k
90dxGO	7.850 ^{ab}	12.200 ^a	7.760 ^a	11.136 ^a	0.246 ^a	89.16 ^b

d: days; TO: Tobacco seed oil; SO: Sunflower oil; GO: Groundnut oil; Means having different superscripts within a column are significantly different at 1%

than auto-oxidation from 30 days of storage on wards with maximum increase at 90 days of storage (Table 3). Interaction between storage periods and oils was significant. Generally, it was observed that for each period of storage, tobacco seed oil and sunflower oil were at a par and both were significantly higher than groundnut oil (Table 4). Interaction between oxidation methods and oils was significant. pAV in tobacco seed oil and

sunflower oil were at a par for each method of oxidation and was significantly higher than groundnut oil in both oxidation methods (Table 5). p-AV is a measure of secondary reaction products that occur during lipid oxidation and it is more robust indicator of the quality of oil and the extent of oxidation. A higher level of p-AV for oil samples exposed to sunlight when compared to stored at ambient may be attributed to the high rate of formation of

Table 5 : Interaction between oxidation methods (f2) and oil types (f3) on chemical changes during storage

Factors	Peroxide value (meqkg ⁻¹)	p-Anisidine value	Conjugated trienes (ε ^{1%} λ.268)	Conjugated dienes (ε ^{1%} λ.232)	Free fatty acids (%)	Iodine value (g l 100 g ⁻¹)
AOxTO	4.678 ^b	6.606 ^b	2.693 ^c	5.569 ^c	0.143 ^b	130.3 ^a
AOxSO	3.663 ^d	6.03 ^b	2.380 ^d	4.778 ^d	0.140 ^b	119.2 ^b
AOxGO	3.049 ^e	4.950 ^d	1.937 ^e	3.986 ^e	0.119 ^c	94.9 ^f
POxTO	5.433 ^a	7.940 ^a	3.790 ^a	6.825 ^a	0.161 ^a	129.2 ^a
POxSO	4.920 ^a	8.320 ^a	3.395 ^b	6.236 ^b	0.159 ^a	118.5 ^b
POxGO	4.114 ^c	5.770 ^c	2.612 ^{cd}	4.860 ^d	0.141 ^b	93.1 ^e

AO: Auto-oxidation; PO: Photo-oxidation; TO: Tobacco seed oil; SO: Sunflower oil; GO: Groundnut oil; Means having different superscripts within a column are significantly different at 1%

secondary oxidation products in the former. Formation of higher levels of secondary oxidation products in oils exposed to light has been reported (Khan and Shahidi, 2002).

Estimation of CD and CT is a good measure of oxidative state of oils (McGinley, 1991). Conjugated trienes content increased with increase in storage period in both oxidation methods in three oils. The analysis of variance for trienes showed significant difference among the means of storage period, oxidation methods and oils at 1% level of significance. ANOVA for trienes explained 99.5% variation and the coefficient of variation of 10.6% which was reasonably small (Table 1). The average of three storage periods 0, 15, 30 days were at a par, whereas remaining four storage periods viz., 45, 60, 75 and 90 were significantly different among themselves and also with the earlier storage periods (Table 2). Trienes gradually and significantly increased from 45 to 90 days. Triene content was significantly higher in photo-oxidation (3.266) than auto-oxidation (2.337). Three oils were significantly different among themselves. Tobacco seed oil showed significantly higher content of trienes (3.24) over sunflower (2.89) and groundnut oils (2.27) (Table 2). Marcela *et al.* (2011) reported an increase in conjugated dienes and trienes increased significantly throughout two months storage period of walnut oil.

Interaction of storage periods and methods of oxidation for trienes were significant at 1% level. The first three storage periods with the methods of oxidations were at a par. From 45 days storage onwards, showed significant variation at each storage period between photo and auto-oxidations. Interaction between storage periods and oils was significant. Treatment combinations at 0, 15, 30 days of storage with three oils were not significant. Among most of treatment combinations, tobacco seed oil was at a par with sunflower oil (Table 4). Interaction between oxidation methods and oils was significant. Average content of trienes in auto-oxidation of tobacco seed oil (2.69) and sunflower oil (2.38) were at a par while both of them were significantly higher than groundnut oil (1.94) (Table 5). Similar trend was observed with photo-oxidation. Tobacco seed oil showed 11.62 and 10.42% higher levels of trienes as compared to sunflower oil in auto and

photo-oxidations, respectively. Grujić Slavica *et al.* (2011) reported an increase in peroxide value, p-anisidine of the auto- and photooxidized sunflower oil samples and confirmed the decrease in oil quality during storage for 3 months. The PV of virgin olive oil samples increased progressively during the 6 to 8 week storage indicating a greater primary oxidation with a relatively high level of hydroperoxides (Xueqi *et al.*, 2014).

Analysis of variance for conjugated dienes explained 99% variation and the co-efficient of variation was 9%, which was very small. Conjugated dienes content increased with increase in storage period in both photo and auto-oxidations. Interaction between the storage period, oxidation methods and oils was not significant (Table 1). Zero days of storage showed average conjugated dienes of 0.03, which was significantly lower than the rest of dates and the highest was 13.38 at 90 days of storage. All the dates of storage were significant except 15 and 30 days. Dienes increased significantly from 45 days onwards with the higher values at 75 and 90 days of storage. Conjugated dienes content in auto-oxidation (4.78) was significantly less than photo-oxidation (5.98). Tobacco oil had the highest conjugated dienes (6.20) followed by sunflower (5.52) and groundnut oils (4.42) and which were significant among themselves (Table 2).

Interaction between storage periods and oxidation methods was significant. Conjugated diene content was significantly varied from 45 day of storage period onwards between the oxidation methods at each of storage period and photo-oxidation was significantly higher than auto-oxidation (Table 3). Interaction between storage periods and oils was significant. The treatment combination up to 15 days of storage with three oils was at a par. Treatment combination from 30 days of storage with three oils ranging from 1.59 to 14.87 was significant. For each period of storage from 30 days onwards, tobacco seed oil was significantly higher followed by sunflower oil and groundnut oil (Table 4). Among oils, as the storage period increased from 30 days onwards, there was a significant increase in dienes. Treatment combinations for oxidation methods and oils were significant. All three oils were significantly different in auto-oxidation while in photo-oxidation tobacco seed oil and sunflower

oil were at par and were significantly higher than groundnut oil. A higher value of conjugated dienes and conjugated trienes in light-exposed oil samples as compared with those stored at ambient might be attributed to the accelerated rate of photo-oxidation and exposure to sunlight radiations that might speed up the oxidative alterations (Pan *et al.*, 2004). Increase in conjugated dienes and conjugated trienes were observed in sunflower oil and soybean oils exposed to different storage conditions (Anwar *et al.*, 2007).

The co-efficient of variation in FFA values was 9.57 and ANOVA has explained 97.6% of variation. Interaction between storage period and oxidation methods were highly significant, whereas rest of the interactions viz., storage period x oils, and oxidation methods x oils were not significant (Table 1).

Free fatty acids content at 0, 15 and 30 days of storage were at a par, whereas 30 and 45 days were also at a par. Free fatty acids significantly increased from 45 days of storage onwards reaching a maximum (0.278%) at 90 days (Table 2). Free fatty acids content in photo-oxidation (0.154%) was significantly higher than auto-oxidation (0.134%). Tobacco seed oil (0.143%) and sunflower oil (0.140%) were at a par and significantly higher than groundnut oil. Both oxidation methods were at a par for different storage periods up to 60 days and were significantly different at 75 and 90 days of storage (Table 2). The PV of virgin olive oil samples increased progressively during the 6 to 8 week storage indicating a greater primary oxidation with a relatively high level of hydroperoxides.

Natural fats and oils are triglyceride form when freshly extracted from the source. On prolonged storage, the triglycerides begin to break down giving rise to free fatty acids due to the presence of moisture in the oil, elevated temperature and, most important of all, lipases (enzyme) coming from the source or contaminating microorganisms (Chandrasekharan, 2013). Presence of excess free fatty acids is an indicator to unnatural state of oil. The present study reveals that the FFA content in tobacco seed oil was at a par with sunflower oil and higher than groundnut oil. Anwar *et al.* (2007) reported increased FFA content due to oxidation in soybean oil stored at ambient and sun light conditions. Xueqi *et al.* (2014) reported significant increase in free fatty acids of sunflower oil stored at 25°C during the storage period.

The initial iodine values (IV) were 135.2, 123.7 and 98 g l 100 g⁻¹ in tobacco seed oil, sunflower oil and groundnut oil respectively. The co-efficient of variation for iodine value was 2.87 which were small indicating small variation among the iodine values. The ANOVA explained 97% of variability (Table 1). The iodine values decreased from 118.78 to 107.78 g l 100 g⁻¹ with increase in storage period from 0 to 90 days (Table 2). Only zero days of storage were significantly different with 60, 75 and 90 days, while 15 and 30 days of storage were significant with 75 and 90 days. Three oils were significantly different for iodine values.

Tobacco seed oil showed significantly higher value (129.78 g l 100g⁻¹) followed by sunflower oil (118.88 g l 100g⁻¹) and groundnut oil (94.05 g l 100g⁻¹) (Table 2). The other treatment combinations were not significant. Iodine value is an indicator of unsaturation and tobacco seed oil showed higher IV indicating the higher level of unsaturated fatty acids. Decrease in IV is an indicator of lipid oxidation (Naz *et al.*, 2004). Tobacco seed oil and sunflower oil showed similar decreasing trends in IVs with increase in storage period.

In the present study, oxidative changes of tobacco seed oil showed nearly similar trends with that of sunflower oil with respect to pAV and FFA, and a marginal increase in conjugated dienes and conjugated trienes in auto-oxidation. Low PV even at 90 days of storage in the present studies indicates that the tobacco seed oil is of good quality. The study suggests that even though tobacco seed oil contains higher levels of unsaturated fatty acids it can be stored like any other edible oils.

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